

A SIMPLE STEREOSELECTIVE SYNTHESIS OF (1*R*,3*R*,5*S*)-1,3-DIMETHYL-2,9-DIOXA-  
BICYCLO[3.3.1]NONANE USING REGIOSELECTIVE RING-OPENING OF  
(*R*)- $\beta$ -METHYL- $\beta$ -PROPIOLACTONE

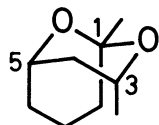
Toshio SATO, Toshiyuki ITOH, Chihiro HATTORI, and Tamotsu FUJISAWA\*  
Chemistry Department of Resources, Mie University, Tsu, Mie 514

A facile stereoselective synthesis of (1*R*,3*R*,5*S*)-1,3-dimethyl-2,9-dioxabicyclo[3.3.1]nonane was achieved from (*R*)-2-hydroxy-7-octen-4-one, which was easily prepared by the copper-catalyzed reaction of (*R*)- $\beta$ -methyl- $\beta$ -propiolactone with vinylmagnesium bromide.

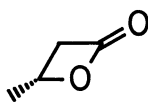
*Endo*-1,3-dimethyl-2,9-dioxabicyclo[3.3.1]nonane (1) is an interesting biologically active substance isolated from Norway spruce infested by a major timber pest, the ambrosia beetle (*Trypodendron lineatum* Oliv.), and has shown to exhibit an important role for causing host selection of this beetle.<sup>1,2)</sup> Although four optically active isomers of *endo*- and *exo*-1 have been recently synthesized from D-glucose in somewhat lengthy steps,<sup>3)</sup> the absolute configuration of natural 1 has been unknown yet. We wish to report here a fairly simple stereoselective way to synthesize an *endo*-enantiomer, [1*R*,3*R*,5*S*]-isomer using the regioselective ring opening reaction of (*R*)- $\beta$ -methyl- $\beta$ -propiolactone (2).

Previously, the S<sub>N</sub>2 type copper-catalyzed reaction of optically active  $\beta$ -methyl- $\beta$ -propiolactone with Grignard reagents, accompanying the ring-fission between the  $\beta$ -carbon and the ether oxygen, has been shown to provide an efficient method for the synthesis of optically active natural products *via* chiral 3-substituted butyric acid derivatives.<sup>4)</sup> On the contrary, in the present synthesis the regioselective reaction of lactone 2 with vinylmagnesium bromide at the acyl carbon is utilized for the synthesis of the key starting material, *i.e.*, (*R*)-2-hydroxy-7-octen-4-one (3), which seems to be formed by the Michael addition of vinylmagnesium bromide to the initial ring-opening product, (*R*)-5-hydroxy-1-hexen-3-one.

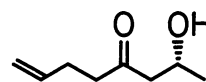
The lactone 2 was easily prepared<sup>4)</sup> from (*S*)-(+)-3-bromobutyric acid (98% ee; [M]<sub>546</sub><sup>25</sup>+114.2° (c 2.3, 2M HClO<sub>4</sub>), lit.<sup>5)</sup> [M]<sub>546</sub><sup>25</sup>+116.5°) in a yield of 70%; [ $\alpha$ ]<sub>D</sub><sup>22</sup>+28.8° (c 4.3, CHCl<sub>3</sub>) (95% ee).<sup>6)</sup> When vinylmagnesium bromide (2 eq) was added to a mixture of 2 and copper(I) iodide (2 mol%) in THF-Me<sub>2</sub>S (20:1) at -10 °C and the reaction mixture was stirred for 1 h at the same temperature, the desired (*R*)-



1



2

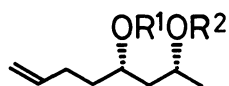


3

hydroxy ketone<sup>7)</sup> **3** was isolated in a high yield of 89% by distillation;  $[\alpha]_D^{22} -57.2^\circ$  (c 2.27,  $\text{CHCl}_3$ ); bp  $80^\circ\text{C}/0.3$  mmHg (kugelrohr); NMR ( $\text{CCl}_4$ )  $\delta$  1.15 (3H, d,  $J = 6$  Hz), 2.3 - 2.5 (6H, m), 3.7 (1H, s), 4.0 - 4.4 (1H, m), and 4.8 - 6.0 (3H, m).

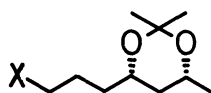
Next key step for construction of the *endo*-isomer **1** was stereoselective reduction of **3** to *erythro*-7-octene-2,4-diol (**4**), which was easily achieved by the procedure reported by Narasaka and Pai.<sup>8)</sup> Thus, treatment of **3** with tributylborane and subsequent reduction with  $\text{NaBH}_4$  furnished *erythro*-diol **4** (*erythro:threo* = 94:6). *Erythro*-**4** was isolated by silica gel TLC (AcOEt:hexane = 2:1) in 85% yield;  $[\alpha]_D^{22} -15.0^\circ$  (c 1.14, MeOH).<sup>9)</sup>

Treatment of **4** with 2-methoxypropene gave acetonide (**5**) (93%); bp  $95^\circ\text{C}/15$  mmHg (kugelrohr);  $[\alpha]_D^{24} -12.05^\circ$  (c 1.33,  $\text{CHCl}_3$ ). The acetonide was converted to alcohol (**6**) by hydroboration (90%), and oxidized by pyridinium dichromate (PDC) in  $\text{CH}_2\text{Cl}_2$  to give aldehyde (**7**) (60%). Treatment of **7** with methylmagnesium bromide in THF at  $0^\circ\text{C}$  furnished secondary alcohol (**8**) (quant.), which was oxidized by PDC in DMF to ketone (**9**) in 86% yield. Ketone **9** was smoothly converted to bicyclononane **1** by the reported procedure;<sup>3)</sup> bp  $160^\circ\text{C}/120$  mmHg (Kugelrohr);  $[\alpha]_D^{24} -32.4^\circ$  (c 0.25, pentane), lit.<sup>3)</sup>  $[\alpha]_D^{27} -37.3^\circ$ .



**4**:  $R^1, R^2 = \text{H}$

**5**:  $R^1, R^2 = \begin{matrix} \text{Me} \\ \diagup \\ \text{C} \\ \diagdown \\ \text{Me} \end{matrix}$



**6**:  $X = \text{CH}_2\text{OH}$

**7**:  $X = \text{CHO}$

**8**:  $X = \text{CH}(\text{OH})\text{CH}_3$

**9**:  $X = \text{COCH}_3$

#### References

- 1) V. Heemann and W. Francke, *Naturwissenschaften*, **63**, 344 (1976); *Planta Med.*, **32**, 342 (1977) [*Chem. Abstr.*, **88**, 101563h (1978)].
- 2) H. Gerlach and P. Künzler, *Helv. Chim. Acta*, **60**, 638 (1977).
- 3) H. Redlich, B. Schneider, R. W. Hoffmann, and K. J. Geueke, *Justus Liebigs Ann. Chem.*, **1983**, 393.
- 4) T. Sato, T. Kawara, A. Nishizawa, and T. Fujisawa, *Tetrahedron Lett.*, **21**, 3377 (1980); T. Fujisawa, T. Sato, T. Kawara, and K. Ohashi, *ibid.*, **22**, 4823 (1981); T. Sato, K. Naruse, and T. Fujisawa, *ibid.*, **23**, 3587 (1982).
- 5) A. R. Olson and R. J. Miller, *J. Am. Chem. Soc.*, **60**, 2687 (1938).
- 6) Although optical rotation of (*S*)- $\beta$ -methyl- $\beta$ -propiolactone was determined recently by NMR spectroscopy as  $\alpha_D - 27.8 \pm 1.6^\circ$  (neat, dm) [A. Leborgne, M. Moreau, and N. Spassky, *Tetrahedron Lett.*, **24**, 1027 (1983)], the optical purity of lactone **2** was determined by the optical rotation of (*R*)-(-)-ethyl 3-hydroxybutyrate obtained by the reaction of lactone **2** with ethyl alcohol in the presence of a catalytic amount of sodium ethoxide;  $[\alpha]_D^{24} -40.7^\circ$  (c 1.09,  $\text{CHCl}_3$ ), for the (*S*)-isomer, lit.  $[\alpha]_D^{25} +43^\circ$  (c 0.93,  $\text{CHCl}_3$ ) [H. Hungerbühler, D. Seebach, and D. Wasmuth, *Helv. Chim. Acta*, **64**, 1467 (1981)].
- 7) The absolute configuration of **3** was tentatively assigned as *R* by considering the reaction mechanism of the nucleophilic attack to the acyl carbon.
- 8) K. Narasaka and H. C. Pai, *Chem. Lett.*, **1980**, 1415.
- 9) H. Gerlach and H. Wetter, *Helv. Chim. Acta*, **57**, 2306 (1974); P. A. Bartlett and K. K. Jernstedt, *Tetrahedron Lett.*, **21**, 1607 (1980).

(Received June 27, 1983)